2017 Photosynthesis and Respiration STQ MS

2017 / H2 / AJC PRELIM / P2 Q6

1 In anaerobic respiration in yeast, the pyruvate molecules are broken down to produce ethanol and carbon dioxide. The release of carbon dioxide can be used to investigate the rate of anaerobic respiration.

Fig. 6.1 shows and experiment which was set up to find the rate of anaerobic respiration.



Fig. 6.1

The meniscus moves down the tube as carbon dioxide is released.

Table 6.1 shows the distance moved by the meniscus from the start point. This was recorded every 10 minutes.

Tab	le	6.	1

Time/ min	0	10	20	30	40	50	60	70	80	90
Distance travelled by meniscus from start point/ mm	0	1	2	5	9	14	21	45	73	98

(a) The rate of anaerobic respiration can be calculated by using the rate of movement of the meniscus.

Calculate the rate of anaerobic respiration between 70 and 80 minutes.

You will lose marks if you do not show your working.								
Rate of Anaerobic Respiration	= (73-45) mm / (80-70) min							
	= 28 mm / 10 min							
	= 2.8 mm min ⁻¹							

(b) This experiment was repeated three more times. Each time, the glucose (a monosaccharide) was replaced with a different disaccharide sugar:

Maltose – a disaccharide of glucose and glucose Sucrose – a disaccharide of glucose and fructose Lactose – a disaccharide of glucose and galactose.

Tables 6.2 (a), (b) and (c) show the results of these experiments.

[2

Time/ min Distance travelled by	0	10	20	30	40	50	60	70	80	90
meniscus from start point/ mm	0	0	0	0	0	2	3	6	9	12
	Т	able 6.2	2 (b): Us	sing suc	crose					
Time/ min	0	10	20	30	40	50	60	70	80	90
meniscus from start point/ mm	0	0	0	1	3	11	22	37	48	61
	Т	able 6.	2 (c): U	sing lac	tose					
Time/ min	0	10	20	30	40	50	60	70	80	90
meniscus from start point/ mm	0	0	0	0	0	0	0	0	0	0

With reference to the information provided in Tables 6.2 (a), (b) and (c) and your biological knowledge:

(i) Describe the difference in the results for maltose and sucrose, and suggest one explanation for this difference,

Glucose is the respiratory substrate for glycolysis;

Maltose: meniscus only starts moving at 50 min vs sucrose at 30 min/ movement of sucrose more than maltose + data; Enzyme to break down sucrose is more readily available/at higher concentration than enzymes [2 to break down maltose;

[2

(ii Suggest two explanations for the results for lactose.

) Yeast does not have proteins channels that allows uptake of lacotse; Yeast does not encode for lactase that breaks down lactose to glucose and galactose; AVP;

(c) An electron micrograph of yeast, *Candida albicans*, is shown in Fig. 6.2.



Fig. 6.2

- (i) On Fig. 6.2, label site of
 - i. Glycolysis
 - ii. Oxidative phosphorylation
- (ii State one visible structure of mitochondria from Fig. 6.2 and describe how it supports mitochondria's fur)ction.

Highly folded inner membrane + Increase surface area for embedment of more ETC/ ATP synthase to increase rate of ATP production Membrane bound/ double membrane + Enclose matrix that contains enzymes for link reaction [1 and Kreb's cycle/ compartmentalizes matrix for optimum conditions for enzymes to work;

(ii Besides location, compare between oxidative phosphorylation and photophosphorylation.

)

Features	Oxidative phosphorylation	Photophosphorylation
i editires	Oxidative phosphorylation	i notopnosphorylation
Functions in the presence of	oxygen	light
Source of energy		light
	NADH and FADH2	
No. of electron transport chain	1	2
Electron flow	linear – one-way	linear or cyclic
Final electron acceptor	oxygen	NADP (non-cyclic)
		PSI reaction center (cyclic)
Involvement of water	water produced	photolysis of water
Establishment of proton gradient	protons pumped <u>outwards</u> from <u>matrix</u> across <u>inner</u> <u>mitochondrial membrane</u> into <u>intermembrane space</u>	protons pumped <u>inwards</u> from <u>stroma</u> across <u>thylakoid</u> <u>membrane</u> into <u>thylakoid</u> <u>space</u>
Products	ATP, water	ATP, NADPH, oxygen

[Total: 13 marks]

2017 / H2 / CJC PRELIM / P2 Q6

Microalgae have been extensively studied for various purposes, such as the production of biomass as a source of valuable chemicals of health foods and for wastewater treatment. Recently, microalgal photosynthesis was considered to be an effective means to reduce the emission of carbon dioxide, a major greenhouse gas, in the atmosphere. Light is the most important factor affecting microalgal photosynthesis kinetics. In general, most microalgal mass culture systems are limited by light, because light is easily absorbed and scattered by the microalgal cells. Therefore, understanding and quantification of light dependence of microalgal activity is of great importance in designing an efficient photobioreactor, in predicting process performance, and in optimizing operating conditions. [2

[4

(a) Explain the trends seen when red, green and daylight (at 0.123gL-1) are compared.

ANS [L2] Novel

[5]

- 1. Comparing at 1500µEm²s⁻¹ red light yields the highest Photosynthetic rate 5.5 mgL-1h-1 due to the presence of P680 and P700 absorbing best at those wavelengths.
- 2. Comparing at 1500µEm⁻²s⁻¹ Green light yields 3.0 mgL-1h-1 the lowest absorption as green light is reflected.
- Comparing at 1500µEm⁻²s⁻¹ daylight yields a moderate Photosynthetic rate 4.5 mgL-1h-1 due to the presence of P680 and P700 but subject to the efficiency of light capturing.
- 4. General trend within graph, at low light intensity, all three showed that Light was a Limiting Factor.
- 5. At higher intensities there is a plateau in all three graphs where light is no longer a limiting factor.
- Fig. 6.2 shows a schematic showing the functional relationship between light harvesting complexes (LHC) and photosystems II & I. Regulatory complexes are also shown comprising of kinases and the regulation of excess energy between PS II and I.



Gollan et al 2015 Photosynthetic light reactions: integral to chloroplast retrograde signalling. Current Opinion in Plant Biology 27:180-191 modified.

Fig. 6.2

(b) Explain what is the LHC and its role in photosynthesis.

ANS [L1] Novel

[2]

- 1. LHC is the Light Harvesting Complex comprising mainly of **Carotenoids and special chlorophyll**;
- 2. Role is to consolidate / channel energy to the photosystems in this way helps in the promotion of electrons within Photosystems;
- (c) With reference to Fig. 6.2 explain the role of electrons in the photosynthesis as they move from Photosystem II to Photosystem I.

- 1. electrons are of a higher energy state once promoted, are then passed down the ETC where energy lost in the transfer is used to pump protons into the thylakoid space;
- 2. Chemiosmosis of H+ then drives the synthesis of ATP using ATP synthase. ATP will then be used in the Calvin cycle.
- 3. Electrons passed from PSII through the ETC reach and replenish PSI.
- (d) With reference to Fig. 6.2 suggest the implications of the role of LHC and PSII core protein phosphorylation from Photosystem II to Photosystem I.

ANS [L3] Novel

[3]

- 1. LHC and PSII core protein help with the **distribution of energy between PSII and PSI.**
- 2. At high light intensity there is a redistribution of energy so that bleaching does not occur.
- 3. At low light intensities, there is a channeling of energy so that photosynthesis will continue.

[Total: 13]

2017 / H2 / DHS PRELIM / P2 Q8

Question 8

(a) (i)

Dehydrogenase reduces NAD+ to NADH

when isocitrate is converted to α -ketoglutarate / succinyl-CoA; OR when malate is converted to oxaloacetate.

NADH carries the electron and proton to electron transport chain for ATP synthesis via oxidative phosphorylation.

OR

Dehydrogenase reduces FAD^{2+} to $FADH_2$

when succinate is converted to fumerate.

FADH₂ carries the electron and proton to electron transport chain for ATP synthesis via oxidative phosphorylation.

(a)(ii)

Depth B

Mean dehydrogenase activity was lower, ranging from 1.5-2.5AU, while that of depth A was higher, ranging from approximately 3.2-5.5AU.

At a greater depth, oxygen concentration is lower. Hence, rate of oxidative phosphorylation is lower.

Regeneration of NAD⁺ / FAD²⁺ is slower hence there is lesser substrates for effective collision.

(b)

lactic acid fermentation

Pyruvate converted to lactate, NADH oxidised to NAD.

Carbon dioxide concentration

(c)(ii)

Carbon dioxide concentration is higher near factories More CO₂ for fixation with Ribulose Bisphosphate

2017 / H2 / JJC PRELIM / P2 Q6

4 Fig. 6.1 shows some stages in mammalian respiration.



Fig. 6.1

- (a) Name the processes taking place during Stage D and state precisely where they occur. [3]
- 1. Link reaction mitochondrial matrix;;
- 2. Krebs cycle mitochondrial matrix;;
- 3. Oxidative phosphorylation inner mitochondrial membrane;;
- (b) Intermediates produced at the end of Stages B and C are important in the conversion of carbohydrates to lipids such as triglycerides. Some of the triose phosphate can be converted into glycerol-3-phosphate, while pyruvate can undergo further reactions to form intermediates required for the synthesis of fatty acids.
 - (i) Describe the formation of triglycerides. [3]
- 1. A triglyceride is formed by condensation reactions between <u>1 glycerol and</u> <u>3 fatty acids;</u>;
- 2. Each of <u>glycerol's hydroxyl/–OH groups</u> condenses with the <u>carboxyl/</u> <u>–COOH group of a fatty acid;;</u>
- 3. In each <u>condensation reaction</u>, one <u>water molecule is removed</u>, resulting in the formation of an <u>ester bond/linkage;</u>;

- (ii) State two roles of triglycerides in living organisms. [2]
- 1. Triglycerides serve as a good energy source;;
- 2. Triglycerides are a weight efficient means for organisms to store energy / serve as good energy storage molecules;;
- 3. Triglycerides serve as a good source of metabolic water;;
- 4. Triglycerides are good thermal insulators that reduce / prevent excessive heat loss from the body;;
- 5. Provide buoyancy to aquatic animals;; (any 2)
- (c) The first reaction in Stage A is catalysed by the enzyme hexokinase. It has been observed that hexokinase is bound to the outer mitochondrial membrane in muscle cells which undergo high rates of glycolysis.



Fig. 6.2

With reference to the role of mitochondria and Fig. 6.2, suggest how the association of hexokinase with mitochondria can lead to high rates of glycolysis. [2]

- 1. Mitochondria are the site of aerobic respiration to synthesise ATP;;
- 2. Due to the <u>close proximity</u> of hexokinase to the mitochondria, <u>ATP produced</u> by the mitochondria <u>can easily be used</u> by hexokinase <u>to phosphorylate glucose</u>;; increasing the rate of glycolysis.

Fig. 6.3 shows an electron micrograph of a mitochondrion.



Fig. 6.3

- (d) With reference to features visible in Fig. 6.3, outline how the structure of the mitochondrion is adapted for its function. [2]
- 1. The inner mitochondrial membrane is <u>highly folded</u>, providing <u>a large surface area</u> where stalked particles, enzymes and electron carriers of the electron transport chain (ETC) *(any 1 e.g.)* needed for <u>aerobic respiration</u> can be located;;
- 2. The mitochondrion is enclosed by <u>double membranes</u> separated by (an extremely narrow fluid-filled space) intermembrane <u>space</u>, allowing for <u>compartmentalisation</u> within the mitochondrion / specialised metabolic pathways to take place in different areas;;

[Total: 12]

- 1 2
- 3
- 4
- 4
- 5
- 6 Heart muscle cells and epidermal cells were extracted from Chinese hamsters. The cells were lysed and the mitochondria and cytosol were isolated. The mitochondria and cytosol were then mixed and re-suspended in a culture of essential nutrients. This suspension system was used to study the process of cellular respiration.

At time 0, glucose was added to the system. At Time X, Digitonin, a detergent which disrupts membranes was introduced to the suspension system. A probe was used to measure the concentrations of ATP as well as the pH level in the mitochondria.

The experimental results are recorded in the graphs shown. **Fig. 8.1** shows the rate of ATP production for heart muscle cells and epidermal cells. **Fig. 8.2** shows the pH level of the mitochondria in both heart muscle and epidermal cells.



 (a) Account for the difference in the level of ATP production in both tissues after glucose was added. Heart cells produce a <u>higher</u> level of ATP as it contains <u>more mitochondria</u> than epidermal cells; (b) With reference to **Fig. 8.1**, explain the changes in ATP production over time for the heart muscle cell suspension.

	2 marks max for the first 3 marking points	
	High energy electrons from reduced coenzymes are passed down a series of electron carriers on the ETC, each with an energy level lower than the one preceding it;	
	Energy from the <u>flow of electrons</u> is used to <u>actively pump</u> protons from matrix to <u>intermembrane space</u> , through conformational change of proteins in ETC;	
	Rate of ATP production increases with time before addition of X due to protons <u>diffuse</u> down the <u>electrochemical proton gradient</u> back into mitochondrial matrix through <u>ATP synthase/ stalked particles</u> , synthesizing ATP from ADP and Pi;	
	Ref to effect resulting from the membrane damage e.g. no ETC / ATP synthase / electrochemical gradient etc	
	Initial lag in ATP production because glycolysis is occurring to produce 2 ATP per glucose by substrate level phosphorylation; OR	
	Rate of ATP production levels off because glycolysis can still occur and decreases to zero when glucose is used up;	
		[3]
(c)	With reference to Fig. 8.2 , state which region of the mitochondrion the pH probe was Explain your conclusion.	measuring.
	Intermembrane space;	
	Ref to pH4 + protons are actively pumped from matrix into intermembrane space;	
		[2]
(d)	Suggest why cytosol was used to re-suspend the mitochondria.	
	Enzymes involved in glycolysis are present in the cytosol;	
		[1]
(e)	From your biological knowledge, explain the adaptation of the double membrane for it production of energy.	s role in the
	Membrane is impermeable to protons, creating electrochemical proton gradient across the inner mitochondrial membrane;	
	Highly folded inner membrane increase surface area for stalked particles containing ATP synthase and electron carriers to be embedded;	
	<u>Compartmentalisation</u> so that reactions can occur in different locations (ref. to provision of optimal conditions for enzymes such as that for Krebs' Cycle to work);	
		[2] [Total: 9]

2017 / H2 / PJC PRELIM / P2 Q7

Question 7 [12 marks]

An experiment was conducted to investigate how various factors affect the rate of photosynthesis in cabbage. **Fig. 7.1** below shows the results of the experiments conducted.



Fig. 7.1

- (a) With reference to Fig. 7.1,
 - (i) state the best conditions for the growth of cabbage. [1]

5% carbon dioxide, 25°C, and (any value between 8.8-10) lux;;

- (ii) explain the region marked Y. [2]
- a. Light intensity is no longer the limiting factor / light saturation is achieved, ;
- b. as mean mass of cabbage plants <u>remains constant at 305g</u> (accept range between 300-310g) beyond 8 lux (accept range between 7.6-8) lux;
- c. Mean mass of cabbage plants will only increase if <u>temperature is increased from 15°C to</u> <u>25°C</u>. ;
- d. Rate of photosynthesis is limited by temperature;
- (iii) describe and explain the effect of increasing carbon dioxide concentration on the mean mass of cabbage at 25 °C. [3]
- a. As the carbon dioxide concentration <u>increases</u> from <u>0.03% to 5%</u>, the maximum mass <u>increases</u> from <u>70g to 370g</u> of cabbage;;

Max 1.5 for marking points b) to e)

- b. More CO₂ is used for more carbon fixation during light independent reaction;
- c. Increase in carbon dioxide concentration will <u>increase the frequency of effective</u> <u>collisions</u> between enzyme, <u>Rubisco</u> and substrates, <u>RuBP and CO₂</u>;
- d. Hence rate of enzyme-substrate complex formation increases;
- e. Rate of formation of <u>glyceraldehyde-3-phosphate</u> increases;
- R: product, need to mention which product is responsible for the increase in mass

f. More glyceraldehyde-3-phosphate molecules are converted to form more glucose and Pyruvater of plycelysis needs to move from the cytosol into the mitochondrion. After some processing, pyruvate will be converted to acetyl Co A which then enters the Kreb cycle.

Compare Graph B and D only. Idea of increase/more and rate/per unit time should be Fighenuties at hele time increase in the should be be a state of the should be a state of the should be be a state

R: description of light-dependent reactions which does not explain why mean mass of cabbage increases

- (iv) The average carbon dioxide content of the natural environment is 0.035%. Using this fact, and the information given in **Fig. 7.1**, what conclusion can be made about how carbon dioxide affects rate of photosynthesis in the natural environment? [2]
- a. .035% in the natural environment is close to 0.03% in the experiment,
- b. shows that carbon dioxide concentration is a <u>limiting factor</u> on the rate of photosynthesis <u>in the natural environment;</u>;
- c. At maximum light intensity at <u>25°C</u>, Graph D showed higher rate of photosynthesis which resulted in <u>370g</u> mass of cabbage whereas
- d. Graph B showed lower rate of photosynthesis which resulted in only <u>70g</u> of cabbage;; OR
- e. At maximum light intensity at <u>15°C</u>, Graph A showed higher rate of photosynthesis which resulted in <u>30g</u> mass of cabbage whereas
- f. Graph C showed lower rate of photosynthesis which resulted in only <u>305g</u> (accept range between 300-310) of cabbage;;
- (b) While photosynthesis is the process by which carbon dioxide and water are used as starting materials for the synthesis of glucose using light energy, respiration involves releasing chemical energy in organic molecules such as glucose by oxidation and made available to living cells in the form of ATP. In particular, the yield of ATP under aerobic and anaerobic respiration are very different.

Explain the small yield of ATP from anaerobic respiration in both yeast and animals. [4]

- a. During anaerobic respiration, there is no oxygen, thus there is no <u>final electron acceptor</u> for oxidative phosphorylation to take place;
- b. This will result in electron flow via electron transport chain being blocked;
- c. When this happens, there will be <u>no protons pumped into intermembrane space from the</u> <u>matrix of mitochondria;</u>
- d. Proton gradient (and hence proton motive force) will be dissipated;
- e. No chemiosmosis via ATP synthase \rightarrow no ATP production via oxidative phosphorylation;
- f. NAD+ and FAD will not be regenerated so link reaction and Krebs cycle cannot take place to generate more ATP, NADH and FADH₂;
- g. Only <u>glycolysis</u> can take place to generate a <u>net gain of 2 ATP molecules per glucose</u> <u>molecule;</u>
- h. Anaerobic respiration via lactate fermentation or alcoholic fermentation does not produce ATP but only regenerates NAD+ for glycolysis to continue;
- i. only a net of 2 ATP molecules are synthesized via glycolysis via <u>substrate level</u> <u>phosphorylation;</u>
- j. a large amount of energy is locked in lactate / alcohol;

2017 / H2 / RI PRELIM / P2 Q6

6

(a) Name the region within the mitochondrion where the Krebs cycle occurs and using the symbol X, indicate this region on Fig. 6.1. [1]

X: mitochondrial matrix

(b) Pyruvate requires the help of pyruvate translocase to enter the mitochondrion. The structure of pyruvate is shown in Fig. 6.2.



Fig. 6.2

Pyruvate translocase is located on the membrane labelled 'A' in Fig. 6.1.

Fig. 6.3 shows how pyruvate is transported across the membrane A.



 (i) Explain why pyruvate requires the help of pyruvate translocase to cross membrane A. [2]
 1. Pyruvate is <u>charged*/polar*</u> and therefore <u>hydrophilic*</u> OR

pyruvate is too large to enter through transient pores;

- hence requires a <u>specific</u>* <u>transport protein</u> / tpt protein provides a <u>hydrophilic pore /</u> <u>channel;</u>
- 3. to cross inner membrane of mitochondrion which has a *hydrophobic core**;
- (ii) With reference to Fig. 6.3, describe the transport of pyruvate across membrane A. [1] pyruvate transport is <u>coupled</u> with \underline{H}^{\pm} in the <u>same direction</u> and moved from

- (c) Other than pyruvate translocase, the electron transport chain and ATP synthase can also be found on membrane A.
 - (i) Explain how pyruvate translocase, the electron transport chain and ATP synthase are held in membrane A. [2]
 - <u>Non-polar</u> <u>hydrocarbon chains</u> of phospholipid bilayer can form <u>hydrophobic</u> <u>interactions</u> with <u>non-polar R groups</u> of amino acid residues found on exterior surface of proteins;
 - 2. <u>Charged phosphate head</u> of phospholipid bilayer/aqueous solutions on either side of membrane can <u>interact</u> (e.g hydrogen bond, ionic bond) with <u>charged /polar R groups</u> of amino acid residues found on exterior surface of proteins.
 - (ii) Explain how mitochondria are adapted to contain many structures such as pyruvate translocase, the electron transport chain and ATP synthase. [1] Inner membrane/cristae (membrane A) is highly folded to increase surface area for placement of the proteins;
 - $H^{+} H^{+} H^{+$
- (d) Fig. 6.4 shows a section of membrane A.

electron transport chain

Fig. 6.4

Dinitrophenol (DNP), cyanide and oligomycin are chemicals that interfere with the normal functioning of the components shown in membrane A.

(i) DNP can shuttle H⁺ across biological membranes.

With reference to Fig. 6.4, explain why fewer ATP molecules were produced when DNP was added to mitochondria. [1] Lower proton gradient \rightarrow <u>no/fewer H</u>[±] moves <u>down concentration gradient</u> through <u>ATP</u> synthase \rightarrow fewer ATP molecules produced;

(ii) Cyanide inhibits complex IV shown in Fig. 6.4.

Explain why in the presence of cyanide, oxygen consumption decreases and lactate production increases.[4]

- 1. Complex IV will <u>no longer</u> be able to <u>pass electrons to O₂</u> the final electron acceptor, resulting in less O₂ consumption;
- <u>Electron carriers remain reduced/prevention of reoxidation of electron carriers</u>, thus preventing flow of electrons down <u>electron transport chain</u>* (ETC);
- 3. NADH and FADH₂ will not be able to donate electrons to ETC, therefore <u>no</u> <u>regeneration of NAD</u>[±] and FAD*;

- 4. *lactate dehydrogenase** will reduce <u>pyruvate to lactate</u> thus increasing lactate production;
- NADH is reoxidised to form <u>NAD</u>^{+*}, so that <u>glycolysis can continue</u> to generate more ATP;
- (iii) Oligomycin inhibits ATP synthase by blocking the proton channel, as seen in Fig. 6.4.

Indicate in the graph below, the effect of administering a fixed amount of oligomycin on the rate of ATP synthesis. [1]

Rate without oligomycin with oligomycin [1] Concentration of ADP and P_i

[Total: 13]

2017 / H2 / RVHS PRELIM / P2 Q8



	In mito	an investigation to det ochondria were incubated	termine the effering in four ways:	ect of chemica	al M on respira	tion,						
		1. with glucose										
		2. with pyruvate										
		3. with glucose and chemical M										
		4. with pyruvate and chemical M										
	The results are summarised in Table 8.1. Table 8.1											
	CO2 evolution CO2 by oxidative consumption N N											
		Glucose	x	x	x							
		Pyruvate	✓	✓	✓							
		Glucose + chemical M	x	x	x							
		Pyruvate + chemical M	✓	\checkmark	x							
(c)	(i)	i) Explain why carbon dioxide is produced when mitochondria are incubated with pyruvate but not when incubated with glucose.										
	 no glycolytic enzymes in mitochondria; glycolysis does not occur in the mitochondria / glycolysis can only occur in the cytosol; glucose cannot be oxidised to form pyruvate; pyruvate can enter mitochondria but glucose cannot; CO₂ produced by decarboxylation in link reaction; 											
	(ii)	Suggest why when r chemical M , oxygen co	mitochondria are onsumption occu	e incubated wit irs but not ATP p	h pyruvate and production.	[2]						
		1. Chemical M only of ADP/no flow ATP synthase);;	/ block ATP syı of H⁺ down co	nthase so no p ncentration gra	hosphorylation adient (through							
		2. Chemical M doo oxygen;;	es not affect l	ETC to transfe	er electrons to							

2017 / H2 / SAJC PRELIM / P2 Q5

A student set up an experiment to investigate the effect of carbon dioxide on photosynthesis. First, he de-starched a small potted plant by putting it in the dark for two days. Then, he chose two leaves and inserted them into conical flasks, **A** and **B**, fitted with rubber stoppers. Lithium hydroxide was placed in Flask **B** to absorb all carbon dioxide present. The plant was then left under a table lamp for 15 minutes. Fig 5.1 shows the experimental setup.



Fig 5.1

He removed a sample of each leaf every 5 minutes (by punching out a leaf disc of approximately 0.2 cm in diameter, using a single-hole puncher) and return the leaves to their respective flasks immediately. Each leaf disc was then tested for the presence of ribulose bisphosphate (RuBP) and starch. Table 5.2 shows the results he obtained.

				lč	adie 5.2			
Flask	Conce	ntration of	f RuBP / µr	nolm-²	Concentration of starch / µmolm-2			
	0 min	5 min	10 min	15 min	0 min	5 min	10 min	15 min
Α	0.0	2.2	3.0	3.1	0.0	2.1	3.4	6.5
В	0.0	2.7	4.2	6.8	0.0	0.3	0.5	0.5

(a) State two other variables which must be kept constant to maximize the validity of the results obtained for this experiment.

.....[1]

- 1 Size of the leaves chosen (in flask A and flask B)
- 2 Distance from light source/light intensity
- 3 Temperature
- 4 Size of flask (affecting the volume of gas)

[Reject: leaves must be from the same plant as diagram already show only one pot of plants; pH of soil as fluctuations in pH will affect both leaves equally and hence, not affect the results]

(b) With reference to Table 5.2, describe the relationship between the presence of carbon dioxide and concentration of starch.

.....[2]

- 1 In the presence of carbon dioxide (i.e. Flask A), starch concentration increased from <u>0.0 to</u> <u> 6.5μ molm²</u> from <u>0 - 15 min</u>.
- 2 while in the absence of carbon dioxide (i.e. Flask B), starch concentration increased much **less**, from <u>0.0 to 0.5μ molm²</u> from <u>0 15 min</u>.

OR

- 1 The presence of carbon dioxide leads to a higher concentration of starch
- 2 E.g. at 15min, starch concentration in the presence of carbon dioxide (i.e. Flask A) is at <u>6.5 µmolm²</u> while_in the absence of carbon dioxide (i.e. Flask B), starch concentration is at <u>0.5µmolm²</u>

(c) Explain the absence of RuBP in both leaves at the start of the experiment.

-[2]
 - 1 In the dark (during de-starching), the leaves carried out carbon fixation to convert **RuBP to PGA**.
 - 2 However, in the dark, no <u>ATP</u> and <u>NADPH</u> were produced,
 - 3 hence, chloroplasts are unable to convert **PGA to PGAL** and to regenerate **RuBP from PGAL**.
- (d) The increase in RuBP concentration for the leaf in Flask A reached a plateau from 10 min to 15 min of exposure to light but continued to increase in the leaf in Flask B up to 15 min. Explain why.

.....[3]

- 1 (In Flask A, concentration of RuBP reaches a plateau because) the RuBP that is being used up (in carbon fixation) equals the RuBP that is regenerated (from PGAL in the Calvin cycle).
- 2 (In Flask B, exposure to light causes) the production of <u>ATP</u> and <u>NADPH</u> in the lightdependent reactions, which will convert existing PGA to PGAL (don't double penalize from part (c)), and **regenerate RuBP from PGAL**.
- 3 However, in the absence of carbon dioxide, carbon fixation does not occur, and RuBP **accumulates** in the leaf.

The student watered the potted plant too excessively, causing the soil to become waterlogged. Fortunately, the roots of this plant could carry out anaerobic respiration under low oxygen conditions in the soil.

(e) Outline the process of anaerobic respiration in the roots under waterlogged conditions.

.....[4]

- 1 Glucose is converted to pyruvate via <u>glycolysis</u> in the <u>cytosol</u>.
- 2 Pyruvate is then decarboxylated to acetaldehyde/ethanal with the release of <u>CO₂</u>.
- 3 This reaction is catalysed by pyruvate decarboxylase.
- 4 NADH then reduces acetaldehyde/ethanal to ethanol,
- 5 catalysed by <u>alcohol dehydrogenase</u>.
- 6 NAD+ is **regenerated** for use in glycolysis to generate some **ATP** continuously.

[Q5: 12 marks]

2017 / H2 / TJC PRELIM / P2 Q7

9 Glucose and fructose are two common fruit sugars used in winemaking. Another sugar used in the fermentation industry is sucrose. The effects of the three sugars on fermentation by yeast were investigated and the results are shown in Fig. 7.1 and Fig. 7.2.



(a Describe how ethanol is formed by yeast. [2]

- 1. <u>Pyruvate is first decarboxylated to ethanal</u>. The enzyme <u>is pyruvate</u> <u>decarboxylase</u>.
- 2. Ethanal is then <u>reduced by NADH to form ethanol</u>. <u>NAD+</u> is <u>regenerated</u>. The enzyme involved is <u>alcohol dehydrogenase</u>.

With reference to Fig. 7.1 and Fig. 7.2, explain the order in which the sugars were utilized by yeast for fermentation. [4]

- 1. Glucose, then sucrose, followed by fructose.
- Concentration of glucose <u>decreased at a faster rate</u> than concentration of fructose – at day 2, concentration of glucose dropped from 110 g/L to 50 g/L, as compared to concentration of fructose which dropped from 115 g/L to 90 g/L
- 3. <u>CO₂ produced was also higher</u> at 11 mL for glucose than for fructose (3 mL) and sucrose (6 mL)
- 4. These show that <u>more glucose was used</u> for fermentation, resulting in a higher level of CO₂ produced during the <u>conversion of pyruvate</u> to ethanal

Fig. 7.3 shows a respirometer.

- 1. Potassium hydroxide absorbs CO₂ produced
- 2. O_2 will be absorbed by the seeds
- 3. Thus, the level of the coloured oil will move up towards tube B

Suggest how the compensation point of a plant will be affected when it undergoes anaerobic respiration.

- 1. Compensation point is the <u>amount of light intensity</u> when <u>rate of</u> <u>respiration corresponds with rate of respiration.</u>
- 2. During <u>anaerobic respiration, less CO₂</u> is produced as compared to aerobic respiration.
- 3. Compensation point will <u>likely decrease</u>

[Total: 11]