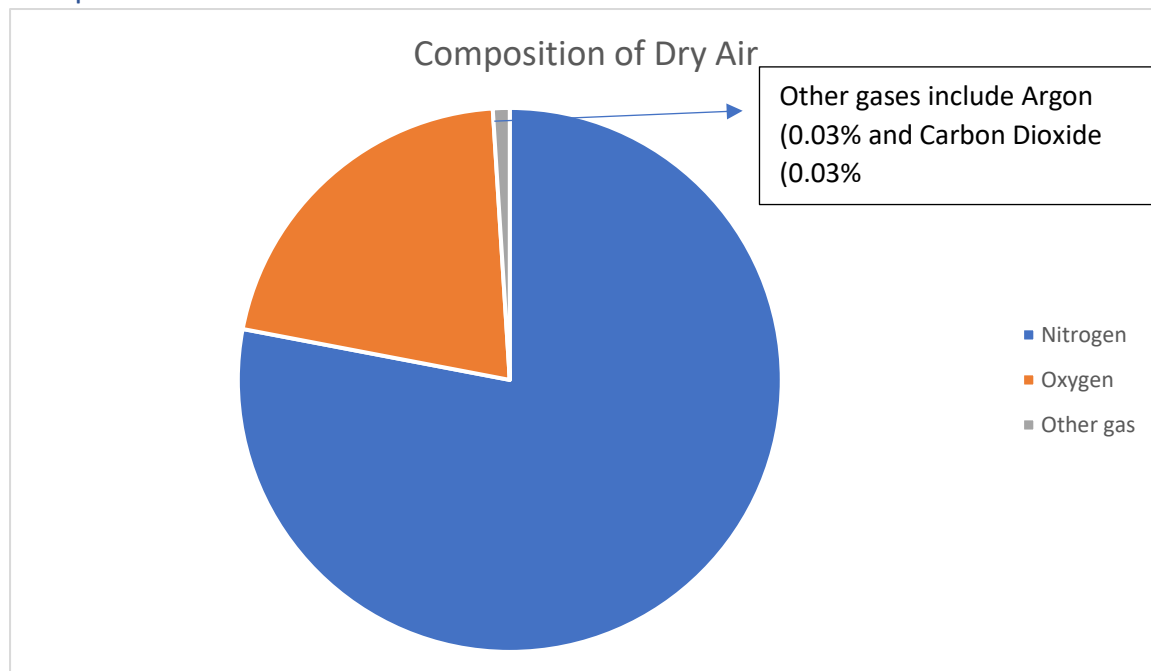


Air and the Atmosphere

Content



Composition of Air



Air Pollution

Pollutant	Sources	Effects
Carbon monoxide, CO	<ul style="list-style-type: none">• Incomplete combustion of fuels	<ul style="list-style-type: none">• Fatigue, headache, death• CO binds irreversibly with haemoglobin in blood, preventing it from carrying oxygen.• $2\text{CH}_4 + 3\text{O}_2 \rightarrow 2\text{CO} + 4\text{H}_2\text{O}$
Sulfur dioxide, SO ₂	<ul style="list-style-type: none">• Volcanic eruptions• Burning of fossil fuels containing sulfur	<ul style="list-style-type: none">• Irritation to eyes of lungs• Acid rain (corrodes limestone buildings)• $2\text{SO}_2 + 2\text{H}_2\text{O} + \text{O}_2 \rightarrow 2\text{H}_2\text{SO}_4$• $4\text{NO}_2 + 2\text{H}_2\text{O} + \text{O}_2 \rightarrow 4\text{HNO}_3$
Nitrous dioxide, NO ₂ Nitrous oxide, NO	<ul style="list-style-type: none">• Lightning activity• Internal combustion in engines• Strong $\text{N} \equiv \text{N}$ bonds so combustion only occurs at high temps	
Unburnt hydrocarbons	<ul style="list-style-type: none">• Incomplete combustion of fuels	<ul style="list-style-type: none">• Reacts with NO₂ to form ozone
Methane, CH ₄	<ul style="list-style-type: none">• Decaying matter• Digestion of food	<ul style="list-style-type: none">• Greenhouse gas effect
Ozone, O ₃	<ul style="list-style-type: none">• Reaction of unburnt hydrocarbons with NO₂	<ul style="list-style-type: none">• Photochemical smog

Catalytic Converter

- Made up of Rhodium, Platinum, Palladium

List of reactions

1	$CO + NO \rightarrow N_2 + CO_2$
2	$CO + O_2 \rightarrow 2CO_2$
3	$2NO_2 \rightarrow N_2 + O_2$
4	$C_xH_{4x} + 2xO_2 \rightarrow CO_2 + 2xH_2O$

Flue Gas Desulfurization

- Waste gases from combustion of fossil fuels are called flue gases
- Desulfurization is the process of removing sulfur dioxide from flue gases
- $SO_2 + CaCO_3 \rightarrow CaSO_3 + CO_2$
- $2CaSO_3 + O_2 \rightarrow 2CaSO_4$
- $SO_2 + CaO \rightarrow CaSO_3$

Depletion of the ozone layer

- Caused by chlorofluorocarbons in aerosols and refrigeration coolants
- In UV rays, CFCs decompose to form chlorine atoms
- React with ozone to form chlorine monoxide and oxygen
- Ozone layer is destroyed and UV rays can reach earth

Carbon Cycle

Respiration $\rightarrow C_6H_{12}O_6 + 6O_2 \rightarrow 6CO_2 + 6H_2O$

Photosynthesis $\rightarrow 6CO_2 + 6H_2O \rightarrow C_6H_{12}O_6 + 6O_2$

Greenhouse effect

- Increased concentrations of greenhouse gases due to anthropogenic activities
 - o Enhanced greenhouse effect \rightarrow rise in temperature
- Consequences?
 - More drought, flooding
 - Less snow + ice
 - More extreme weather incidents
 - RISING SEA LEVEL
 - Warmer oceans + partially melted glaciers + other ice \rightarrow increase sea level. Ocean water expands when warm \rightarrow further rise in sea level